

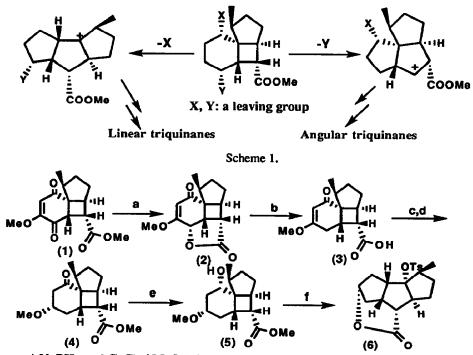
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## Facile Synthesis of Both Linear and Angular Triquinanes from The Tricyclo[5.4.0.0<sup>1,5</sup>]undec-9-ene-8,11-dione Efficiently Produced by a Combination of Electro- and Photochemical Reactions

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Abstract: Both linear and angular triquinanes have been efficiently synthesized from 9-methoxy- $6\alpha$ -methoxycarbonyl-2 $\beta$ -methyltricyclo[5.4.0.0<sup>1,5</sup>]undec-9-ene-8,11-dione.

In the course of our synthetic study on bioactive substances by means of electrochemical methodology, 1) we could synthesize 9-methoxy- $6\alpha$ -methoxycarbonyl- $2\beta$ -methyltricyclo[5.4.0.0<sup>1,5</sup>]undec-9-ene-8,11-dione (1)<sup>2</sup>) in high yield. This compound 1 is regarded as the common synthetic intermediate for both linear and angular triquinane-type compounds, which are quite attractive because of their structural and biological properties,<sup>3</sup>) as seen in Scheme 1.

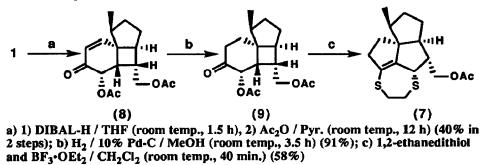


a) NaBH<sub>4</sub> and CeCl<sub>3</sub> / MeOH (room temp., 2 h)(70%); b) H<sub>2</sub> / Pd-C / EtOAc (room temp., 5.5 h)(90%); c) MeI and  $K_2CO_3$  / DMF (room temp., 13 h) (93%); d) H<sub>2</sub> / Pd-C / EtOAc (room temp., 3.5 h)(64%); e) NaBH<sub>4</sub> / MeOH (room temp., 40 min.)(79%); f) p-TsOH / toluene (refluxing temp., 15 min.) (60%)

Scheme 2.

On NaBH4 reduction in the presence of CeCl3, the tricyclic compound 1 was readily converted into the lactone  $2^{(4)}$  in 70% yield, which was further subjected to catalytic hydrogenation to afford  $3^{(4)}$  in 90% yield. After esterification using MeI and K2CO3, 3 was hydrogenated again over catalytic amounts of 10% Pd-C to give the corresponding keto-ester (4),<sup>4</sup> in good yield, which was treated with NaBH4 in MeOH to give 5,<sup>4</sup> in high yield. When treated with p-TsOH in toluene at refluxing temperature for 15 min., 5 was selectively converted into the desired linear triquinane-type compound  $6^{5}$  in 60% yield, as seen in Scheme 2. We could also convert the tricyclic compound 1 into the angular triguinane-type compound 7 in short step, as follows.

The tricyclic compound 1 was treated with diisobutylaluminium hydride in THF followed by acetylation resulting in removal of the acetoxyl group at C11-position leading to the formation of the  $\alpha_{\beta}$ -unsaturated ketone  $8.^{(4)}$  in 40% overall yield, which was subjected to catalytic hydrogenation using 10% Pd-C in EtOAc to afford the corresponding saturated ketone 9,4) in high yield. When treated with 1,2-ethanedithiol and excess BF3•OEt2 (1:10) in CH<sub>2</sub>Cl<sub>2</sub> at room temperature, 9 was readily converted into the desired angular triquinane-type compound  $7^{(5)}$  in 58% yield, as seen in Scheme 3.



## Scheme 3.

The present study indicates a short synthetic pathway leading to both linear and angular triguinane-type sesquiterpenes. Further synthetic studies on a number of highly oxygenated triquinanes are in progress.

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## REFERENCES

- 1. Yamamura, S.; Shizuri, Y.; Shigemori, H.; Okuno, M.; Ohkubo, M. Tetrahedron 1991, 47, 635. Yamamura, S. "Natural Products Syntheses by Means of Electrochemical Methodology" in "Electroorganic Synthesis: Festschrift for Manual M. Baizer" ed by R. D. Little and M. L. Weinberg,
- Marcel Dekker, Inc., New York (1989), pp. 309-315.
  Maki, S.; Kosemura, S.; Yamamura, S.; Ohba, S. *Tetrahedron Lett.*, 1993, 34, 6083.
  Hudlicky, T.; Rulin, F.; Lovelace, T. C.; Reed, J. W. "Synthesis of Natural Products Containing Five-membered Rings. An Evolution of General Methodology," in "Studies of Natural Products Chemistry," ed by Atta-ur-Rahman, Elsevier Science Publishers B. V., Amsterdam (1989), vol. 3, pp. 3-72 and many references cited therein.
- 4. The spectral data for the new compounds are in accord with the structures assigned, and the stereochemistry of each compound was elucidated by NOE experiments.
- 5. 6: C13H16O2 [m/z 204.1138 (M<sup>+</sup> TsOH)]; IR (film) 1740 cm<sup>-1</sup>;  $\delta$  (CDCl3) 0.87 (3H, d, J = 6.4 Hz), 1.16 (1H, m), 1.25 - 1.34 (2H, complex), 1.52 - 1.83 (4H, complex), 2.02 (1H, dd, J = 2.0, 15.6 Hz), 2.27 (1H, dd, J = 2.9, 14.2 Hz), 2.45 (3H, s), 2.50 (1H, m), 2.59 - 2.67 (2H, complex), 2.73 (1H, br. s), 4.47 (1H, br. s), 7.34 (2H, d, J = 8.3 Hz), and 7.80 (2H, d, J = 8.3 Hz); 7: C17H24O2S2 [m/z 324.1244 (M<sup>+</sup>)]; IR (film) 1740 cm<sup>-1</sup>;  $\delta$  (CDCl<sub>3</sub>) 0.89 (3H, d, J = 5.9 Hz), 1.27 (1H, dd, J = 2.4, 4.9 Hz), 1.30 (1H, dd, J = complex), 2.55 (1H, d, J = 8.8 Hz), 3.05 - 3.20 (5H, complex), 3.92 (1H, dd, J = 7.8, 11.2 H2.4, 4.4 Hz), 1.40 - 1.55 (6H, complex), 1.80 - 1.87 (2H, complex), 2.03 (3H, s), 2.05 - 2.43 (5H, z), and 4.26 (1H, dd, J = 7.3, 11.2 Hz).